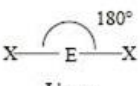
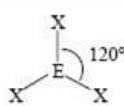
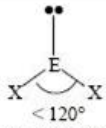
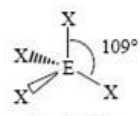
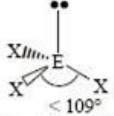

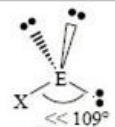


Molecular Models

VSEPR Geometries				
Steric No.	Basic Geometry 0 lone pair	1 lone pair	2 lone pairs	3 lone pairs
2	 Linear			
3	 Trigonal Planar	 Bent or Angular		
4	 Tetrahedral	 Trigonal Pyramid	 Bent or Angular	 Linear

	Lewis Diagram	3-D Sketch	Shape	Symmetrical?
N_2				
H_2S				
CO_2				
NH_3				
$SiBr_4$				
SO_4^{2-}				